Components of Cell-free Extracts of Clostridium pasteurianum W5 Required for Acetylene Reduction and  $N_2$  Fixation

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Summary: The same two protein components, molybdoferredoxin and azoferredoxin, are required for acetylene reduction as were shown previously to be required for  $\rm N_2$  fixation. Neither protein component of the  $\rm N_2$ -fixing system catalyzes either of these two reactions without the other. Both protein components are oxygen- and heat-  $(60^{\rm O})$  sensitive and one of them, azoferredoxin, is also cold- $(0^{\rm O})$  labile. Neither molybdoferredoxin nor azoferredoxin is present in extracts of cells grown on  $\rm NH_3$ . No indication of the need for a third component was observed.

The two protein components, molybdoferredoxin (MoFd) and azoferredoxin (AzoFd), of the nitrogenase system of Clostridium pasteurianum W5 have been partially purified from cells grown on  $N_2$  (Mortenson et al., 1967). The same two protein components have been shown to be required for ATP-dependent  $H_2$  evolution, electron-dependent ATP utilization (Mortenson, 1966), electron-independent ATP utilization (Kennedy & Mortenson, 1969; Jeng et al., 1969a) and acetylene reduction (Moustafa & Mortenson, 1967). Similar protein fractions have also been separated from the nitrogenase of Azotobacter vinelandii (Bulen & LeComte, 1966), Azotobacter chroococcum (Kelly, 1969) and soybean nodules (Klucas et al., 1968).

Recently, it was suggested that AzoFd of <u>C</u>. <u>pasteurianum</u> contains two fractions necessary for  $N_2$  fixation and that one of these two fractions is present in cells grown on either  $N_2$  or  $NH_3$  (Taylor, 1969). In addition the latter author stated that MoFd (his  $C_1S_1$  fraction) when added to an extract of  $NH_3$ -grown cells supported acetylene reduction activity but no  $N_2$  fixation. In light of this report we have reevaluated our results and find that as previously reported 1) AzoFd and MoFd are single protein components that are required for both activities, 2) they are found only in cells grown on  $N_2$ ,

3) the addition of MoFd to an extract of NH<sub>3</sub>-grown cells does not give acetylene reduction activity and 4) there is no third component required.

# Materials and Methods

A Cary 14 recording spectrophotometer was used to record the spectrum of AzoFd before and after treatment at  $60^{\circ}$ . Protein was measured by the Biuret method (Gornall et al., 1949). N<sub>2</sub> fixation was assayed manometrically (Mortenson, 1966). An extract of NH<sub>3</sub>-grown cells was added as a source of the supporting enzymes, ATP: acetate phosphotransferase (for the ATP generating system), and H<sub>2</sub>: ferredoxin oxidoreductase (hydrogenase) and ferredoxin (for the electron donating system). In some experiments sodium dithionite (hydrosulfite) was used in place of the reduced Fd generating system (Burns & Bulen, 1965) and substrate levels of ATP (10mM) were used instead of an ATP generating system (Kennedy et al., 1968). Acetylene reduction was measured by the method of Koch & Evans (1966).

All treatments throughout the purification of MoFd and AzoFd were performed under anaerobic conditions. Cells and crude extracts were prepared as previously described (Mortenson et al., 1967). The two protein components of nitrogenase, MoFd and AzoFd, were separated as described by Mortenson et al., (1967). MoFd with a molecular weight of about 160,000 was further purified by the method of Mortenson et al., (1968) and Morris et al., (1969). It was about 95% pure based on a molybdenum content of 2 atoms per molecule and disc gel electrophoresis examination. AzoFd containing 2 iron and 2 acid-labile sulfide groups per molecule was isolated and purified (about 98% pure) and found to have a molecular weight of approximately 39,000 (Moustafa & Mortenson, 1969; Jeng, et al., 1969b).

### **Results**

MoFd and AzoFd have been found only in extracts of cells that actively fix  $N_2$ . When either of these proteins was added to an extract of cells grown on  $NH_3$ , neither  $N_2$  fixing nor acetylene reducing activity was observed (Fig. 1A & B). Similar results were obtained if sodium pyruvate was used as electron

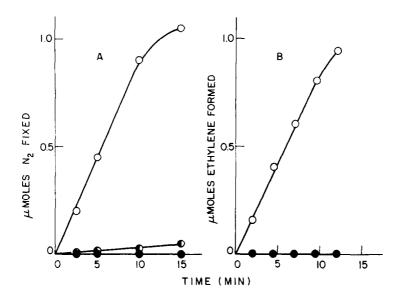


Fig. 1A Requirement of MoFd and AzoFd for  $N_2$  fixation with  $H_2$  as electron donor. The complete system in a total volume of 2.0 ml contained 100  $\mu$ moles of lithium cacodylate buffer at pH 6.75, 10  $\mu$ moles MgCl $_2$ , 19.3 mg protein from extract of cells grown on NH $_3$ , 5 mg MoFd, 1.15 mg AzoFd in the main compartment as well as 5  $\mu$ moles ATP and 50  $\mu$ moles acetyl phosphate in the sidearm. The center well contained 0.1 ml 20% KOH. The gas phase was 0.5 atm  $H_2$  and 0.5 atm  $N_2$  with a control of 0.5 atm A and 0.5 atm  $N_2$ . After 5 min equilibration at 30° under the specified gases, the reactions were started by tipping in the ATP and acetyl phosphate solution from the sidearm. 0-0-0 complete system under  $H_2$  &  $N_2$ ; 0-0-0 complete system minus MoFd and AzoFd under  $H_2$  &  $N_2$ ; 0-0-0 complete system minus MoFd and AzoFd under  $H_2$  &  $N_2$ ; 0-0-0

Fig. 1B Requirement of MoFd and AzoFd for  $C_2H_2$  reduction with  $H_2$  as electron donor. The complete system in a total volume of 2.0 ml contained 100  $\mu$ moles lithium cacodylate buffer at pH 6.75, 10  $\mu$ moles MgCl $_2$ , 8 mg protein from extract of cells grown on NH $_3$ , 5 mg MoFd, 0.3 mg AzoFd, and 20  $\mu$ moles ATP. The gas phase was 0.85 atm  $H_2$  and 0.15 atm  $C_2H_2$ . After 5 min equilibration at 30 under the specified gases, the reactions were started by injecting the ATP solution with a gas tight syringe into the reaction vials. To determine the amount of  $C_2H_2$  reduced, a standard of ethylene was included. 0-0-0 complete system; •-•• complete system minus MoFd and AzoFd or either one of them.

and energy source (Fig. 2). Therefore, both MoFd & AzoFd are required for both activities and neither is found in extracts of cells grown on NH<sub>2</sub>.

Taylor (1969) claimed that AzoFd was heat stable at  $60^{\circ}$ , whereas MoFd was not. We find that when crude N<sub>2</sub>-fixing extracts were treated at  $60^{\circ}$  under H<sub>2</sub> for 10 min., no N<sub>2</sub>-fixing or acetylene reducing activity occurred unless MoFd was added; the total activity was always lower than that of the untreated

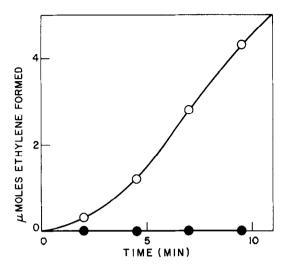


Fig. 2 Requirement of MoFd and AzoFd for  $C_2H_2$  reduction with sodium pyruvate as energy and electron source. The complete system in a total volume of 2.0 ml contained 100  $\mu$ moles lithium cacodylate buffer at pH 6.75, 10  $\mu$ moles MgCl<sub>2</sub>, 50  $\mu$ moles sodium phosphate at the same pH, 19.3 mg protein from extract of cells grown on NH<sub>3</sub>, 5.0 mg MoFd, 1.15 mg AzoFd and 100  $\mu$ moles sodium pyruvate with 0.5  $\mu$ mole of ATP. The gas phase was 0.85 atm A add 0.15 atm  $C_2H_2$ . After 5 min equilibration at 30° under the specified gases, the reactions were started by injecting the pyruvate and ATP solution with a gas tight syringe into the reaction vials. To determine the amount of  $C_2H_2$  reduced, a standard of ethylene was included. 0-0-0 complete system; •-•-• complete system minus MoFd and AzoFd or either one of them.

extract but the specific activity increased (Fig. 3A). However, when purified AzoFd was heated under similar conditions and the heated AzoFd then combined with MoFd, no activity resulted (Fig. 3B). Obviously certain components in the crude extract prevented AzoFd from being heat denatured. Even though no noticeable precipitation occurred when AzoFd was heated to 60°, AzoFd was indeed changed as evidenced by a large change in its spectrum (Fig. 4). Heat treated AzoFd did precipitate readily when added to the assay mixture; some sort of complexing with other components appeared to be necessary to observe denaturation visually. These results show that both MoFd and AzoFd are sensitive to heat (60°) treatment.

As a result of examining fractions of a crude AzoFd preparation collected from a Sephadex G-100 column, Taylor (1969) claimed there probably were two components present in our AzoFd preparation and that both were required for  $N_2$ 

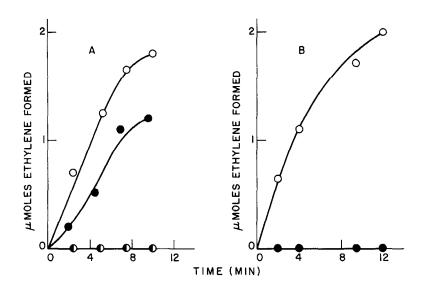


Fig. 3A Heat stability of AzoFd in an extract of cells grown on  $N_2$ . See text for the experimental conditions. The complete system in a total volume of 2.0 ml contained 100  $\mu$ moles lithium cacodylate buffer at pH 6.75, 10  $\mu$ moles MgCl<sub>2</sub>, 0.5 ml of crude extract (21.4 mg/ml) or heat-treated crude extract (10.3 mg/ml after heat treatment) and substrate level of ATP (20  $\mu$ moles). Other conditions were the same as in Fig. 1B. 0-0-0 the original activity of the crude extract; •-•-• the activity of the heat-treated crude extract plus 5.0 mg MoFd; •-•-• the activity of heat-treated crude extract plus 2.0 mg AzoFd or alone.

Fig. 3B Heat stability of purified AzoFd. Purified AzoFd (22 mg/ml) was treated at  $60^{\circ}$  under a H<sub>2</sub> atmosphere for 10 min. The reaction mixture for acetylene reduction in a total volume of 2.0 ml contained 100  $\mu$ moles lithium cacodylate buffer at pH 6.75, 10  $\mu$ moles MgCl<sub>2</sub>; 10  $\mu$ moles sodium dithionite as electron donor, 20  $\mu$ moles ATP, 5.0 mg MoFd and 1.1 mg AzoFd. Other conditions were the same as in Fig. 1B. 0-0-0 the activity in the presence of purified MoFd and AzoFd; •-•-• the activities of MoFd or AzoFd alone and of purified MoFd plus heat-treated AzoFd.

fixation but only one of them for acetylene reduction. To examine this point we subjected purified AzoFd to gel filtration on an anaerobic Sephadex G-100 column (43 x 2.5 cm) with a flow rate of 6.0 ml per hr. Fractions were collected and assayed for acetylene reduction and  $N_2$  fixation. The two activities were not distinguishable, i.e. their ratio was the same in each fraction (Fig. 5). In addition, when fractions from each end of the activity peak were assayed alone and in combination, the activity for  $N_2$  fixation was additive (Table I). There was no stimulation in activity as would be expected if two necessary components had been separated. These data indicate that the same single protein catalyzed both activities.

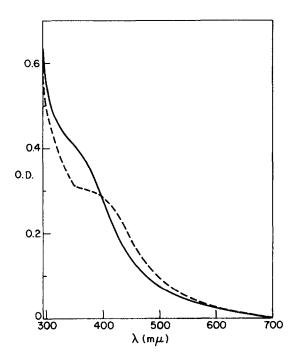


Fig. 4 Visible spectra of native (---) and heat-treated (----) AzoFd under H<sub>2</sub> atmosphere at 25°. 6.5 mg AzoFd in total 3.0 ml was used.

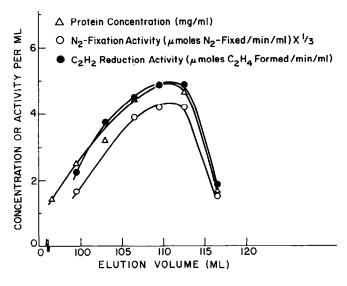


Fig. 5 Activity of fractions of purified AzoFd after chromatography on a Sephadex G-100 column. See text for the protein and activity measurements.

Table I. Demonstration by Sephadex G-100 chromatography of the purity of AzoFd required for  $N_2$  fixation.\*

Fraction #	Amount of Protein Added (mg)	$ m N_2$ Fixation Activity** (m $\mu$ moles $ m N_2$ fixed/min)
#2 (corresponded to the fraction at 100 ml of the elution profile in Fig. 5)	0.26	70
#7 (corresponded to the fraction at 115 ml of the elution profile in Fig. 5)	0.14	50
#2 and #7	0.66***	200 (observed) 190 (calculated)

<sup>\*</sup>The complete system in a total volume of 2.0 ml contained 100  $\mu$ moles lithium cacodylate buffer at pH 6.75, 10  $\mu$ moles MgCl<sub>2</sub>, 4  $\mu$ moles ATP, 50  $\mu$ moles acetyl phosphate, 8.9 mg protein from the extract of cells grown on NH<sub>3</sub>, 4 mg MoFd and indicated amount of AzoFd. Other conditions were the same as in Fig. 1A.

\*\*\*0.52 mg of protein from fraction #2 and 0.14 mg of protein from fraction #7.

## Discussion

Taylor (1969) has described the separation of the nitrogenase complex into at least three necessary protein components. All three fractions ( $\mathbf{C_1S_1}$ ,  $\mathbf{IV_n}$  &  $\mathbf{IV_s}$ ) were required for  $\mathbf{N_2}$  fixation but only two of them ( $\mathbf{C_1S_1}$  &  $\mathbf{IV_s}$ ) were required for acetylene reduction. Fraction  $\mathbf{IV_s}$  was also found in extracts of cells grown on  $\mathbf{NH_3}$ . Both  $\mathbf{IV_n}$  and  $\mathbf{IV_s}$  were stable at  $60^\circ$  under a  $\mathbf{H_2}$  atmosphere for 10 min. MoFd described by Mortenson et al., (1967), must have been a component of Taylor's  $\mathbf{C_1S_1}$  fraction. In contrast to the latter results with crude fractions we find that in addition to MoFd, AzoFd was the only other protein necessary for  $\mathbf{N_2}$  fixation and AzoFd is also required for acetylene reduction. An extract of cells grown on  $\mathbf{NH_3}$ , which according to Taylor should have supported acetylene reduction if MoFd ( $\mathbf{C_1S_1}$ ) was added, was completely inactive when MoFd was added.

The conclusion that only AzoFd is required for the activities of  $\mathrm{N}_2$ 

<sup>\*\*</sup>Calculated from the maximum 5 min rate.

fixation and acetylene reduction in the presence of purified MoFd, an energy source (ATP),  $\mathrm{Mg}^{++}$  and an electron donor (dithionite) is inferred from the following considerations: 1) The purity of our AzoFd has been established from its behavior during ultracentrifugation and disc gel electrophoresis, by a consistent absence of tryptophan from several preparations and by the lack of further separation on Sephadex G-100 (Jeng et al., 1969b); 2) Neither MoFd nor AzoFd was found in extracts of cells grown on  $\mathrm{NH}_3$  even when  $\mathrm{N}_2$  was available; and 3) Only MoFd and AzoFd are required for the activity of  $\mathrm{N}_2$  fixation or acetylene reduction.

The history of the cells grown on  $\mathrm{NH}_3$  is important in the preparation of extracts for this work. Daesch & Mortenson (1968) demonstrated that the turnover rate of nitrogenase in whole cells is extremely low, i.e., it is decreased only by dilution during cell growth in the absence of further nitrogenase biosynthesis when a culture is switched from  $\mathrm{N}_2$ -fixing conditions to an  $\mathrm{NH}_3$ -supplemented medium. Therefore, in order to look for the presence or absence of components of nitrogenase, cells must be grown exponentially for at least 10 generations in the  $\mathrm{NH}_3$ -supplemented medium before preparing the cell-free extract. This could account for Taylor's observations with extracts of  $\mathrm{NH}_3$ -grown cells. Another possible explanation of his results could be that his  $\mathrm{C}_1\mathrm{S}_1$  fraction was contaminated with AzoFd. The different requirements he finds when measuring acetylene reduction versus  $\mathrm{N}_2$  fixation could reflect procedural problems. For example, since the activities reported for acetylene reduction were very low, the measurement of  $\mathrm{N}_2$  fixation under such conditions would be very difficult.

# Acknowledgments

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